

# ③ Dilution - Examples

- Have final concentration & volume  
need to find initial concentration

## Additional calculation questions

1. George weighs 68 kg and requires drug H at a dose of 3 mg/kg per day in four divided doses. Drug H is available as 10 mg capsules. What is the total daily amount of drug H ideally required by George and how many capsules would he take for each dose in practice?

- A 204 mg and 15 capsules
- B 204 mg and 5 capsules
- C 200 mg and 5 capsules
- D 105 mg and 5 capsules
- E 105 mg and 10 capsules

2. YHR pharmaceuticals produces a batch of compressed tablets every fortnight containing 450 mg active ingredient with a mean tablet weight of 0.9 g. Which of the following is the weight of active ingredient that will be required for a total batch size of 6000 kg?

- A 6000 kg
- B 4000 kg
- C 3000 kg
- D 2500 kg
- E 1800 kg

3. You have in your pharmacy an unopened 30 g tube of Locoid Lipocream (hydrocortisone butyrate 0.1%). Which of the following is the amount of Lipobase cream required for diluting 10 g Locoid Lipocream to a concentration of 0.0025% hydrocortisone butyrate?

- A 250 g
- B 390 g**
- C 450 g
- D 575 g
- E 600 g

4. An ointment has the following formula:

Sulphur 4 %w/w

Salicylic acid 10 %w/w

Yellow Soft Paraffin, BP to 100 %w/w

Which of the following are the amounts of sulphur and salicylic acid required to produce 25 g of this ointment?

- A 1 g sulphur and 2.5 g salicylic acid
- B 2.5 g sulphur and 1 g salicylic acid
- C 5 g sulphur and 1 g salicylic acid
- D 8 g sulphur and 8 g salicylic acid
- E 8 g sulphur and 16 g salicylic acid

5. The total amount of Drug G administered to an adult patient weighing 70.0 kg after 10 days' treatment if he requires a single intravenous daily dose of 0.9 mg/kg body weight of Drug G.

- A 0.21 g
- B 0.63 g
- C 0.87 g
- D 1.08 g
- E 1.5 g

6. The weight of nabumetone in 3 capsules of an experimental analgesic formulation with the following formula for 100 capsules:

Nabumetone 50 g

Codeine phosphate 3 g

Lactose 20 g

- A 0.21 g
- B 0.63 g
- C 0.87 g
- D 1.08 g
- E 1.5 g

7. You have in your pharmacy 400 mg potassium permanganate tablets. You are requested to prepare 2 L of a potassium permanganate solution such that the patient will dilute this 1 in 10 to obtain a 0.005% solution suitable for wound washing. How many of these tablets would you dissolve in a small amount of water before making the solution up to a final volume of 2 L with water?

- A 2.5 tablets
- B 3 tablets

$$\text{Dilution factor} = \frac{\text{Initial concn}}{\text{final concn}}$$

$$= \frac{0.1\%}{0.0025\%} = 40$$

∴ original cream must be diluted 1 in 40

- 1 part Locoid lipocream  
- 39 parts Lipobase cream

\* Quantity of lipobase cream required =  $39 \times 10g = 390g$

Final concentration required  
 $0.005\% \text{ w/v} \times 10 = 0.05\% \text{ w/v}$  - dilution factor to make more concentrated  
 $= 0.05g \text{ K}^+ \text{ perm.} / 100ml$   
 • need 2L of 0.05% w/v solution  
 - need to calculate amount in this 2L  
 $0.05g \text{ per } 100ml \times 20 = 1g \text{ per } 2000ml (2L)$   
 (1000mg)  
 - Number of tablets  
 $= 1000 / 400 = 2.5 \text{ tablets}$

(11) - Finding Amount required of active ingredient to make a final solution of certain concentration  
 also has dilution in question

(12) - Volume required of a solution to make another solution at a certain volume & concentration

# dilution examples

- C 3.5 tablets  
 D 4 tablets  
 E 4.5 tablets
8. If 1200 mg potassium permanganate is dissolved in 4 L water, what is the percentage strength of the resulting solution?  
 A 0.003%  
 B 0.015%  
 C 0.03%  
 D 0.15%  
 E 0.3%
9. The recommended intravenous injection dose of gentamicin for the treatment of septicemia for a child aged between 1 month and 12 years is 2.5 mg/kg every 8 h. What volume of gentamicin 40 mg/mL should be given every 8 h to a 10-year-old girl weighing 30 kg?  
 A 2.5 mL  
 B 2.225 mL  
 C 2.0 mL  
 D 1.875 mL  
 E 1.7 mL
10. A 1 in 5000 solution of copper sulphate contains which of the following concentrations?  
 A 50 mg copper sulphate in 500 mL solution  
 B 20 mg copper sulphate in 100 mL solution  
 C 5 mg copper sulphate in 500 mL solution  
 D 100 mg copper sulphate in 1000 mL solution  
 E 50 mg copper sulphate in 300 mL solution

11. Which of the following amounts of chlorhexidine gluconate is required to make 20 mL of a stock solution, such that, when the stock solution is diluted 30 times with water, a final solution of 0.2% w/v chlorhexidine gluconate is produced?

- A 0.2 g  
 B 0.4 g

(11)

## Work backwards

- Final stock solution strength needed:

• 0.2% w/v  
 = 0.2 g chlorhexidine gluconate in 100 mL solution

- multiply by 30 to give the concentration of the original stock solution.

• 6% w/v  
 = 6 g in 100 mL

- finding amounts of chlorhexidine gluconate in the starting w/ 20 mL stock solution.

• 6 g in 100 mL  
~~equation~~  $\frac{20}{100} = 0.2 \times 6 \text{ g}$   
 = 1.2 g of chlorhexidine gluconate.

- C 0.8 g  
 D 1.2 g  
 E 1.8 g

12. Given a 5% w/v solution of tridolan, what volume is required to make 250 mL of a 2% w/v solution?

- A 20 mL  
 B 40 mL  
 C 80 mL  
 D 100 mL  
 E 140 mL

13. The formula for 100 tablets is:

Paracetamol 50 g  
 Codeine hemihydrate 800 mg  
 Caffeine 3 g  
 Lactose 20 g

Which of the following is/are correct?

1. 16 tablets contain 0.128 g codeine hemihydrate.  
 2. A patient taking two tablets four times daily for 3 days ingests 4.8 g lactose from these tablets.  
 3. 32 tablets contain 960 mg caffeine.  
 14. Which of the following is/are correct?  
 1. In order to make 300 g of 2.5% w/w calamine in emulsifying ointment BP, 7.5 g calamine are required.  
 2. 250 mL of a 1.14% w/v solution of povidone-iodine contains 2.85 g of the drug.  
 3. 4 x 250 mg nitrazepam tablets will be required to make 100 mL of a 50 mg/5 mL nitrazepam suspension.

(12) Final Solution

2% w/v solution  
 = 2 g in 100 mL  
 in 250 mL  $2.5 \times 2 \text{ g}$   
 = 5 g in 250 mL

5 g needed

5% w/v tridolan  
 = 5 g in 100 mL  
 • 5 g is needed, therefore we need 100 mL of the 5% w/v tridolan